2004 Air Quality
Data Summary

December 2005
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The 2004 Air Quality Data Summary is available for viewing or downloading on the internet at:

http://www.deq.state.id.us/air/data_reports/monitoring/04_aq_monitoring_report.pdf/

Links to additional documents for download are also available at the web site.

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Introduction

This annual report is issued by the Idaho Department of Environmental Quality (DEQ) to inform the public of air quality throughout Idaho. The purpose of this report is to summarize regional air quality while presenting air monitoring results for six criteria air pollutants (CAPs). The U.S. Environmental Protection Agency (EPA) sets national ambient air quality standards (NAAQS) for these pollutants. These criteria air pollutants are:

- Particulate Matter (PM$_{10} \leq 10$ micrometers, PM$_{2.5} \leq 2.5$ micrometers in diameter)
- Carbon Monoxide
- Sulfur Dioxide
- Nitrogen Dioxide
- Ozone
- Lead

In Idaho, monitoring for the criteria pollutants occurs primarily in areas of high population where the potential for human exposure is greatest. Particulate matter is currently the most common criteria pollutant of concern in Idaho because particulate sources are widespread throughout the state. Common sources include windblown dust, re-entrained road dust, smoke (residential, agricultural, and forest fires), industrial emissions, and motor vehicle emissions. The PM$_{10}$ standard has been in effect since 1987 and it has historically been the particulate of concern. However, PM$_{2.5}$, or PM Fine, has been monitored in Idaho since 1998 and is now the pollutant of concern. Numerous studies have associated fine particulate matter with a variety of respiratory and cardiovascular problems, ranging from aggravated asthma, to irregular heartbeats, heart attacks, and early death in people with heart or lung disease. The PM NAAQS is currently under review by EPA and may be changed in the future.

Another pollutant of concern in Idaho is carbon monoxide (CO). The primary source of CO is incomplete fossil fuel combustion. CO concentrations have the potential to be high in the urbanized areas where automobile traffic is heavy and cars frequently idle at stoplights. The Boise area was the only CO Nonattainment area in the state. When our SIP and Maintenance Plan was accepted by EPA it was reclassified as a Maintenance area on December 27, 2002. No violations of the 1 hour or 8 hour CO NAAQS have occurred since 1991.

Sulfur dioxide (SO$_2$) and nitrogen dioxide (NO$_2$) sources are few and very localized because these pollutants come primarily from large industrial sources (transportation sources also contribute to (NO$_2$). There is little heavy industry in Idaho and elevated SO$_2$ and NO$_2$ concentrations in ambient air are typically not found. However, due to concerns of some localized sources, DEQ has monitored for one or both of these pollutants in Boise, Pocatello, Moyie Springs, Mountain Home and Soda Springs. In the past 10 years of targeted monitoring, DEQ has not measured significant concentrations of these pollutants at these monitoring sites. DEQ initiated NO$_2$ monitoring near Coeur d’Alene in November of 2004 to characterize emissions in the area.

The fifth criteria pollutant monitored by DEQ is ozone. Ozone (O$_3$) has been monitored in the Treasure Valley since 2001. Ozone is created when combustion by-products near the ground react with nitrogen
oxides and other compounds to create photochemical smog. These reactions are stimulated on days of intense sunlight and warm temperatures.

Lead (Pb) is the sixth criteria pollutant and is not currently being monitored by DEQ. Lead (Pb) was monitored in the Shoshone County town of Kellogg, near the Bunker Hill superfund site, because lead was a by-product of the smelting process that occurred in the area for decades. Although a significant problem in the 1970's and early 1980's, airborne Pb concentrations at this monitoring site were very low. DEQ discontinued monitoring for lead in 2002.

Real-time air monitoring data are available on the Internet at http://www.tcsn.net/family/Idaho/index.html. We encourage you to visit our Web site at http://www.deq.state.id.us/ to find more extensive air quality data, educational materials, and discussions of current topics.

We are expanding and refining our Internet site to better serve the residents of Idaho. We want your feedback on our air quality data and program. Please submit your comments via email to Bruce Louks; Monitoring, Modeling, & Emissions Inventory Manager, at Bruce.Louks@deq.idaho.gov or call at 208-373-0294.
Executive Summary for 2004

DEQ continued to monitor Idaho’s air quality in 2004. While Idaho generally enjoys good air quality, in many ways our airsheds are faced with new challenges. Some of these challenges are related to economic and population growth, particularly in terms of vehicles on roadways and growth in new construction.

Criteria air pollutant levels have generally decreased over the last decade to levels below the federal standards due to better control of air pollution; however, two pollutants remain a concern – PM$_{2.5}$ and Ozone. Idaho will need to keep a close watch on PM$_{2.5}$ levels statewide. New PM$_{2.5}$ sites were established in Franklin and Preston at the end of 2004 and beginning of 2005. Ozone levels in the Treasure Valley, while not violating federal standards, are not far below the standard.

DEQ is currently monitoring for toxic air pollutants in the Treasure Valley to determine if concentrations are at levels that could have adverse health affects. These health effects include, but are not limited to, increased cancer risk and respiratory, cardiovascular, and neurological effects. EPA, through their National Air Toxics Assessment (NATA) program, predicts cancer and non-cancer risk values across Idaho’s airsheds using emissions estimates of certain air toxic compounds. Monitoring for these compounds has enabled DEQ to reconcile the NATA predictions.

DEQ is taking action with many partners to face these challenges. These actions include developing emissions reduction strategies, working with planning agencies, and implementing voluntary programs that achieve emissions reductions. Please visit our Web site at http://www.deq.state.id.us/air/prog_issues.cfm for more information about these projects.

There are multiple ways to measure the quality of our air. Some are summarized in the following pages for 2004.
Air Quality Index (AQI)

The AQI is a nationwide index developed by the U.S. EPA that incorporates the criteria pollutants, and is used to report daily air quality. The AQI for Idaho’s counties are summarized below, and discussed in more detail in the section directly following the executive summary.

### 2004 AQI Ratings

<table>
<thead>
<tr>
<th>County</th>
<th># AQI Days</th>
<th>Good (%)</th>
<th>Moderate (%)</th>
<th>Unhealthy for Sensitive Groups (%)</th>
<th>Unhealthy (%)</th>
<th>Highest AQI</th>
</tr>
</thead>
<tbody>
<tr>
<td>Ada</td>
<td>366</td>
<td>87%</td>
<td>13%</td>
<td>0.3%</td>
<td></td>
<td>107</td>
</tr>
<tr>
<td>Bannock</td>
<td>366</td>
<td>93%</td>
<td>7%</td>
<td>0.3%</td>
<td></td>
<td>153</td>
</tr>
<tr>
<td>Boise</td>
<td>357</td>
<td>95%</td>
<td>4%</td>
<td>0.3%</td>
<td></td>
<td>115</td>
</tr>
<tr>
<td>Bonner</td>
<td>366</td>
<td>95%</td>
<td>5%</td>
<td>0.5%</td>
<td></td>
<td>110</td>
</tr>
<tr>
<td>Bonneville</td>
<td>310</td>
<td>97%</td>
<td>2%</td>
<td>0.6%</td>
<td></td>
<td>137</td>
</tr>
<tr>
<td>Butte</td>
<td>351</td>
<td>99%</td>
<td>1%</td>
<td></td>
<td></td>
<td>51</td>
</tr>
<tr>
<td>Canyon</td>
<td>366</td>
<td>88%</td>
<td>11%</td>
<td>1.4%</td>
<td></td>
<td>118</td>
</tr>
<tr>
<td>Caribou</td>
<td>366</td>
<td>97%</td>
<td>3%</td>
<td></td>
<td></td>
<td>70</td>
</tr>
<tr>
<td>Idaho</td>
<td>365</td>
<td>94%</td>
<td>6%</td>
<td></td>
<td></td>
<td>90</td>
</tr>
<tr>
<td>Kootenai</td>
<td>365</td>
<td>84%</td>
<td>16%</td>
<td>0.5%</td>
<td></td>
<td>124</td>
</tr>
<tr>
<td>Latah</td>
<td>355</td>
<td>99%</td>
<td>1%</td>
<td></td>
<td></td>
<td>57</td>
</tr>
<tr>
<td>Lemhi</td>
<td>329</td>
<td>88%</td>
<td>12%</td>
<td>0.3%</td>
<td></td>
<td>101</td>
</tr>
<tr>
<td>Nez Perce</td>
<td>366</td>
<td>98%</td>
<td>2%</td>
<td></td>
<td></td>
<td>61</td>
</tr>
<tr>
<td>Power</td>
<td>364</td>
<td>97%</td>
<td>3%</td>
<td>0.3%</td>
<td></td>
<td>119</td>
</tr>
<tr>
<td>Shoshone</td>
<td>360</td>
<td>71%</td>
<td>27%</td>
<td>1.7%</td>
<td></td>
<td>118</td>
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<tr>
<td>Twin Falls</td>
<td>326</td>
<td>94%</td>
<td>6%</td>
<td></td>
<td></td>
<td>79</td>
</tr>
<tr>
<td>Valley</td>
<td>310</td>
<td>99%</td>
<td>1%</td>
<td></td>
<td></td>
<td>87</td>
</tr>
</tbody>
</table>

The number of “good” AQI days continues to dominate regionally in Idaho; however, there were brief periods when the air quality degraded into “moderate” or “unhealthy for sensitive groups”. The table above shows the AQI breakdown by percentage in each category for the year. In 2004, the highest AQI value of 153 was recorded in Bannock County on January 16 for PM$_{2.5}$. This value was in the unhealthy range.

There were no NAAQS violations in 2004 for any pollutants. In most cases, pollutant concentrations fell well below standards.
Burn Bans

Each day, DEQ measures the concentration of certain air pollutants throughout the state. DEQ may issue a burn ban when concentrations of these air pollutants reach or exceed the health-based standards or limits established by state law or regulation. Concerned citizens may tune in to the news on their local radio or television station to find out if a burn ban has been issued, or log on to DEQ’s Internet web site at http://www.deq.state.id.us/air/aqindex.cfm. DEQ issues a news bulletin to local news media, law enforcement, and fire officials each time a burn ban is imposed. There were a number of voluntary and only two mandatory bans issued in 2004.

Air Emission Inventory

An emission inventory for criteria air pollutants was conducted for the state for calendar year 2002 and is included in this report. 2004 data for large industrial sources was included. An emissions inventory summarizes the quantities of air pollutants reported by large industrial sources, as well as pollutants from other sources estimated using EPA methods. Emission inventory information is presented in detail for carbon monoxide, particulate matter (10 micron diameter), sulfur dioxide, nitrogen dioxide, and VOCs.

The inventory demonstrates that on-road vehicles are the most significant contributors to criteria pollutant emissions in Idaho. Stationary area sources (home heating, small industrial sources, outdoor burning, etc.) are the major contributor of PM$_{10}$ emissions.
Air Quality Index

The AQI is reported according to a 500-point scale for each of the major criteria air pollutants: ozone, particulate matter (both PM$_{2.5}$ and PM$_{10}$), carbon monoxide, nitrogen dioxide, and sulfur dioxide. The “worst denominator” determines the ranking. For example, if an area has a carbon monoxide value of 132 on a given day and all other pollutants are below 50, the AQI for that day would be 132. The AQI scale breaks down into six categories, listed below. Each category has a corresponding color, shown with pollution concentration breakpoints for each category, also shown in the definitions section of this document.

- **0 - 50: Good.** Satisfactory air quality; little or no risk from pollution.
- **51 - 100: Moderate.** Acceptable air quality; potential moderate health concerns for a very small number of people.
- **101 - 150: Unhealthy for Sensitive Groups.** Air quality is acceptable for the general public, but people with health conditions that make them sensitive to a particular pollutant are at greater risk of health problems.
- **151 - 200: Unhealthy.** Everyone may experience some health effects, more serious for members of sensitive groups.
- **201 - 300: Very Unhealthy.** Everyone may experience more serious health effects.
- **301 - 500: Hazardous.** Health risk is at emergency levels. Everyone is likely to be affected.

The AQI is a national index, so the values and colors used to show local air quality and the associated level of health concern will be the same everywhere you go in the U.S.A.

The number of “good” air quality days continues to dominate regionally in Idaho. However, there were brief periods when the air quality degraded into “moderate” or “unhealthy for sensitive groups.” The table presented in the executive summary shows the AQI breakdown by percentage in each category for the year.

The graph on page 7 presents the annual number of “good” days for several Idaho counties. The number of “good” days has remained relatively high over the last few years for each county. Current reduction in the number of “good” days cannot be directly compared with the numbers before 1999, when PM$_{2.5}$ was added to the index and the “unhealthy” category was divided into “unhealthy” and “unhealthy for sensitive groups” or for 2001 in the Treasure Valley with ozone causing a number of moderate days.

Graphs on pages 8 through 14 present all types of days recorded for the seven Idaho counties. The AQI summaries for each county that support the data presented on graphs are located in the Appendix.
Air Quality for Nez Perce County

Days of the Year Percentage

Year


0% 5% 10% 15% 20% 25% 30% 35% 40% 45% 50% 55% 60% 65% 70% 75% 80% 85% 90% 95% 100%

Good  Moderate  Unhealthy for Sensitive Groups  Unhealthy
Monitoring Network

The Idaho monitoring network is a composite of meteorological and pollutant-specific monitoring equipment. DEQ operates most of the monitors while several tribes operate monitors on tribal lands. Data from the network are either collected manually by field staff or sent directly to engineers and scientists through a telemetry network.

The table on the next page presents a summary of the monitoring stations used and parameters monitored during 2004. Some parameters were monitored for only part of the year.

The monitoring network map on page 18 shows monitoring stations that were active in 2004. Monitoring stations are mainly located in high population areas. There are also a few in rural areas. Some sites are selected to focus on the emissions of a single pollutant or group of sources (for example, near a high traffic volume or residential wood burning area). Monitoring siting and monitoring objectives are discussed in the pollutant-specific sections of this report.

Pollutants are measured using methods approved by EPA. In addition, some pollutants of particular interest are measured using more than one method. These additional methods help engineers and scientists to better understand the presence and behavior of these pollutants. The table on the next page lists the methods used for the various pollutants. It is noteworthy that the tapered element oscillating microbalance (TEOM) method (for particulate matter) is continuous. A TEOM measures mass concentrations at pre-set time intervals (e.g. hourly). The TEOM can also be accessed through telemetry for instantaneous PM concentrations. TEOMs enable real-time data interpretation and are discussed further in the particulate matter section of this report. Additional information on measurement methods is available at EPA’s Web site: [http://www.epa.gov/ttn/amtic/](http://www.epa.gov/ttn/amtic/).

In addition to the criteria air pollutants described in this report, air toxics were monitored in 2004 at a Nampa site. For details on air toxics and chemical toxicity, visit the EPA web site at [www.epa.gov/ttn/atw/index.html](http://www.epa.gov/ttn/atw/index.html).

Fine Particulate Monitoring

Particulate (PM$_{10}$) and fine particulate (PM$_{2.5}$) are measured using a variety of methods in Idaho. EPA considers the federal reference method (FRM) to be the most accurate way to determine PM$_{10}$ and PM$_{2.5}$ concentrations. This method involves pulling in air (at a given flow rate) and trapping particles of a certain size (PM$_{10}$ or PM$_{2.5}$) on a pre-weighed filter. The filter is then weighed again and the resulting mass is divided by volume (determined from flow rate and amount of time) to provide concentration. Particles on the filter can be later chemically analyzed and modeled for more information about the sources of particulate matter. Unfortunately, the FRM does not provide continuous or timely information.

DEQ uses the FRM as well as the TEOM continuous method to provide more time-resolved data (i.e. hourly averages).
## Monitoring Network for 2004

<table>
<thead>
<tr>
<th>Name</th>
<th>Location</th>
<th>PM$_{10}$ ref</th>
<th>PM$_{10}$ teom</th>
<th>PM$_{2.5}$ ref</th>
<th>PM$_{2.5}$ teom</th>
<th>O$_3$</th>
<th>SO$_2$</th>
<th>NO$_x$</th>
<th>CO</th>
<th>Met</th>
<th>Tox</th>
</tr>
</thead>
<tbody>
<tr>
<td>Boise</td>
<td>Boise, various locations</td>
<td>⬤</td>
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<tr>
<td>CDA</td>
<td>930 N 15th, Lakes Middle School, Coeur d’Alene</td>
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<td>Emmett</td>
<td>2195 Schiller Road, Emmett</td>
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<td>Franklin</td>
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<td>Grangeville</td>
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<td>North Holms and Pop Kroll, Idaho Falls</td>
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<td>Middleton</td>
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<tr>
<td>Moscow</td>
<td>1025 Plant Sciences Road</td>
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<td>Mtn Home</td>
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</tr>
<tr>
<td>Nampa</td>
<td>Nampa, various locations</td>
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<tr>
<td>P4/Monsanto</td>
<td>Soda Springs</td>
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</tr>
<tr>
<td>Pinehurst</td>
<td>106 Church Street, Pinehurst</td>
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<td>⬤</td>
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<td></td>
<td>⬤</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Rathdrum</td>
<td>Rathdrum Prairie</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>⬤</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Salmon</td>
<td>Salmon, various locations</td>
<td></td>
<td></td>
<td>⬤</td>
<td>⬤</td>
<td>⬤</td>
<td>⬤</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Sandpoint</td>
<td>Sandpoint</td>
<td></td>
<td></td>
<td>⬤</td>
<td>⬤</td>
<td>⬤</td>
<td>⬤</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>St. Maries</td>
<td>St. Maries</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>⬤</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Tilli</td>
<td>251 W. Tilli Road, 20 miles east of Boise</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>⬤</td>
<td>⬤</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Twin Falls</td>
<td>Smith’s Food Store, Twin Falls</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>⬤</td>
<td>⬤</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

### Notes:
- PM$_{10}$ ref: Particulate Matter 10 micrometers (reference)
- PM$_{10}$ teom: Particulate Matter 10 micrometers (teom continuous)
- PM$_{2.5}$ ref: Particulate Matter 2.5 micrometers (reference)
- PM$_{2.5}$ teom: Particulate Matter 2.5 micrometers (teom continuous)
- Tox: Urban Air Toxics
- CO: Carbon Monoxide
- NO$_x$: Nitrogen Oxide
- O$_3$: Ozone (April through October)
- SO$_2$: Sulfur Dioxide
### Monitoring Methods Used in 2004 in Idaho

<table>
<thead>
<tr>
<th>Pollutant Code</th>
<th>Measurement</th>
<th>Method</th>
<th>Units</th>
</tr>
</thead>
<tbody>
<tr>
<td>CO</td>
<td>Carbon Monoxide</td>
<td>Gas Nondispersive Infrared Radiation</td>
<td>Parts per Million</td>
</tr>
<tr>
<td>NOx</td>
<td>Nitrogen Oxides (NOx)</td>
<td>Chemiluminescence</td>
<td>Parts per Million</td>
</tr>
<tr>
<td>O₃</td>
<td>Ozone</td>
<td>UV Absorption</td>
<td>Parts per Million</td>
</tr>
<tr>
<td>PM₁₀ ref</td>
<td>PM₁₀ Reference</td>
<td>Reference - Hi Vol Andersen/ GMW 1200</td>
<td>Micrograms per Cubic Meter</td>
</tr>
<tr>
<td>PM₁₀ teom</td>
<td>PM₁₀ Teom</td>
<td>R&amp;P Mass Transducer</td>
<td>Micrograms per Cubic Meter</td>
</tr>
<tr>
<td>PM₂.₅ ref</td>
<td>PM₂.₅ Reference</td>
<td>Reference—R&amp;P Partisol 2025</td>
<td>Micrograms per Cubic Meter</td>
</tr>
<tr>
<td>PM₂.₅ teom</td>
<td>PM₂.₅ Teom</td>
<td>R&amp;P Mass Transducer</td>
<td>Micrograms per Cubic Meter</td>
</tr>
<tr>
<td>Tox</td>
<td>Urban Air Toxics</td>
<td>Various Methods</td>
<td>Parts per Billion and µg/m³</td>
</tr>
<tr>
<td>SO₂</td>
<td>Sulfur Dioxide</td>
<td>UV Fluorescence</td>
<td>Parts per Million</td>
</tr>
</tbody>
</table>
Impaired Air Quality

Winter Burn Bans

Idaho has a winter impaired air quality program primarily targeting sources of particulate matter from wood stoves and fireplaces. Idaho’s program is generally implemented through local ordinances in those areas that have historically had winter inversion problems. Generally, these ordinances specify that a wood burning ban will be declared whenever DEQ reports an AQI value of 74 or greater for any pollutant measured by an approved monitor and air stagnation conditions are forecasted to continue for at least 24-hours. In some areas, open burning (even if a valid permit has been issued) is prohibited when DEQ reports an AQI value of 60 or higher for any pollutant measured in the city/region/airshed.

The DEQ online Current Air Quality Report lists the air quality in many cities and regions in Idaho. Each report will list the pollutant being monitored, the AQI, and burn restrictions, if any, for the day. Anyone wanting to know if they can burn can go to this site to see what the forecast is for their area.

From November 1 through March 31, when the air quality is forecasted to be poor, DEQ will issue an air quality alert between 2:00 and 4:00 p.m. each day to notify the local community of the following day’s air quality condition forecasted for their airshed. If conditions are expected to be poor for the weekend, alerts will continue to be issued at the regular time on Saturday and Sunday. These alerts will be reported to local media outlets and to others through an email notification list.

Of the six Idaho regions, three report issuing burn bans during 2004. The Pocatello Regional Office issued one burn ban in the Portneuf Valley from January 9 through January 23. There was a mandatory ban on open burning but a voluntary ban residential wood burning. The Coeur d’Alene Regional Office issued one mandatory ban on open burning covering four counties on October 28, which was lifted the next day. The Boise Regional Office issued six voluntary alerts in the Treasure Valley during the wintertime inversion season.

Summer Ozone Alerts

DEQ forecasts pollution conditions for ozone in the Treasure Valley Monday through Friday using pollutant monitoring data and meteorological information. Because ozone needs heat and sunlight to form, it is considered a summertime problem and is only monitored from May 1 through September 30. Ozone pollution can rise to very high levels when the valley experiences hot days with few clouds in the sky. The Treasure Valley tends to see daily ozone levels that begin to rise in the late morning and then peak in the late afternoon and early evening. This phenomenon follows very closely with the time of day that the temperatures are the hottest and the sun is the highest in the sky. Since we have no control over our weather characteristics, we have to focus on controlling what we put into our air. Under yellow alerts, the public is requested to keep vehicles maintained, limit driving, combine trips, comply with open burning restrictions, refuel after 7:00 p.m. and don’t top off the tank, mow lawns in the evening, and don’t use lighter fluid on the barbecue. These alerts will be reported to local media outlets and to others through an email notification list. There were nine yellow alerts for ozone issued in 2004. The plan can be viewed at: http://www.deq.state.id.us/air/data_reports/reports/ada_co/pollution_response_plan_summer.pdf.
Regional Air Emissions Inventory

This section presents an air emissions inventory summary for four of the six criteria pollutants (CO, NOx, PM10, SO2) and volatile organic compounds (VOC), a precursor of ozone. The two criteria pollutants not covered in this section are lead and ozone. Lead is no longer an issue in Idaho and DEQ has only recently started to collect ozone monitoring data. An emissions inventory is useful because it helps to identify the sources of pollutants. Identified sources of pollutants can then be addressed in an effort to reduce emissions through improved technologies, campaigns, community programs, and education to change behaviors, and economic incentives. The National Emission Inventory (NEI) is prepared by EPA every three years, with input from local and state agencies. The inventory discussed below includes data from the 2002 Idaho emission inventory for this region, as well as 2004 large facility data that were collected by DEQ.

Source Categories

There are four general categories that are used to characterize emission sources, and virtually hundreds of subcategories. The four general categories include the following:

- Point Sources
- Mobile On-Road Sources
- Non-Road Sources
- Stationary Area Sources

Point sources are those that many people think of when thinking of air pollution. These include large industries that emit several tons of pollution or more per year from a single location. A description of the thresholds associated with these sources is included in the definitions section of this document.

Mobile on-road sources include cars, trucks, and buses, both commercial and private. This category includes vehicles that run on both gasoline and diesel fuel. As with stationary area sources, on-road mobile sources contribute significantly to air pollution in Idaho.

Non-road sources include, for example, farm vehicles, construction vehicles, aircraft, trains, locomotives, and garden equipment.

Area sources are stationary and typically do not individually emit as much tonnage as point sources. Area sources include commercial businesses such as dry cleaners, printers, and small construction, as well as everyday activities such as burning in a wood stove or fireplace. Although area sources on an individual basis emit far less than point sources, the large numbers of these activities can make them a significant contributor to air pollution in Idaho.

Criteria Air Pollutants

An emission inventory was performed by DEQ for calendar year 2002 that summarized the quantities of criteria air pollutants for the four categories described above. This inventory covers only anthropogenic (man made) sources; biogenic sources of volatile organics such as vegetation and crops are not included.
2004 information is used for the point sources since this is the latest complete inventory year of data. The 2004 point sources are those large facilities with greater than 250 tons per year of any criteria pollutant. The inventory covered the entire state and was based on the following sources of information:

- 2004 large facility annual emission inventory reports (point sources)
- Emission factor and activity level derived estimates (area and mobile sources)
- Mobile source models (on-road and non-road mobile sources)

Point sources are the most straightforward source of information for the emission inventory. Facilities are required to report the tonnage of emissions that they release each year. The area source emission estimates may have more uncertainty associated with them, as they must take into account activity levels (for example, the amount that people perform an activity such as burning wood in a fireplace, or driving to work). Area source information is gathered through surveys, census reports, etc. In addition to the uncertainty associated with activity levels, there is also uncertainty regarding the emission factors themselves. These values are typically developed by EPA, in consultation with state and local air agencies and industry. Additional information on emission factors and how they are derived is available at [http://www.epa.gov/oar/oaqps/efactors.html](http://www.epa.gov/oar/oaqps/efactors.html).

The following table presents the ton per year contributions from each source category for criteria pollutants. For particulate matter, only PM$_{10}$ is listed.

### Idaho 2002/2004 Estimated Criteria Air Pollutant Emission Inventory Summary

<table>
<thead>
<tr>
<th>Source Category</th>
<th>CO</th>
<th>PM$_{10}$</th>
<th>NOx</th>
<th>SO$_2$</th>
<th>VOC</th>
</tr>
</thead>
<tbody>
<tr>
<td>Point Sources (Large Facilities)</td>
<td>21,439</td>
<td>1,168</td>
<td>7,333</td>
<td>19,057</td>
<td>678</td>
</tr>
<tr>
<td>On-Road Mobile Sources</td>
<td>436,906</td>
<td>67,529</td>
<td>43,080</td>
<td>1,575</td>
<td>35,343</td>
</tr>
<tr>
<td>Road Dust</td>
<td></td>
<td>66,548</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Gasoline Vehicles</td>
<td>433,248</td>
<td>472</td>
<td>26,067</td>
<td>1,100</td>
<td>34,606</td>
</tr>
<tr>
<td>Diesel Vehicles</td>
<td>3,658</td>
<td>508</td>
<td>17,014</td>
<td>475</td>
<td>737</td>
</tr>
<tr>
<td>Stationary Area Sources</td>
<td>245,439</td>
<td>86,014</td>
<td>32,113</td>
<td>1,051</td>
<td>120,781</td>
</tr>
<tr>
<td>Outdoor Burning</td>
<td>225,553</td>
<td>15,850</td>
<td>2,717</td>
<td>153</td>
<td>1,310</td>
</tr>
<tr>
<td>Indoor Wood Burning</td>
<td>16,523</td>
<td>2,263</td>
<td>170</td>
<td>26</td>
<td>14,979</td>
</tr>
<tr>
<td>Other Sources</td>
<td>3,363</td>
<td>67,901</td>
<td>29,226</td>
<td>872</td>
<td>104,492</td>
</tr>
<tr>
<td>Non-Road Mobile Sources</td>
<td>132,498</td>
<td>2,580</td>
<td>27,972</td>
<td>480</td>
<td>23,261</td>
</tr>
<tr>
<td>Totals</td>
<td>836,283</td>
<td>157,291</td>
<td>110,498</td>
<td>22,163</td>
<td>180,065</td>
</tr>
</tbody>
</table>

This statewide inventory demonstrates that on-road vehicles are the most significant contributor to criteria pollutant emissions in Idaho. Area sources are the major contributor of PM$_{10}$ emissions. Each pollutant is discussed briefly below, and information in the table above is presented graphically.
Particulate Matter (PM$_{10}$)

Area sources are the largest contributor of PM$_{10}$, with 55% of the contribution shown below. The largest area source contributing subcategories are construction, livestock operations, outdoor burning and agricultural tilling. These four combined contribute over 53% of the contribution of area sources. Another large contributor is from on-road vehicles with 43% of the PM$_{10}$ contribution, mostly from road dust.
Carbon Monoxide (CO)

On-road vehicles are clearly the greatest contributor of CO to the atmosphere in the state at 51.8%. If off-road vehicles and equipment are added in with the on-road vehicles, 68% of the CO is from vehicles and equipment. In the graph below, point sources are barely visible as contributors, as they emit less than 3% of the total CO. The second greatest CO contributor is outdoor burning.
Sulfur Dioxide (SO₂)

The large facility point sources comprise the greatest portion of SO₂ emissions, with 86% of the contribution. All vehicles and motorized equipment account for most of the remaining SO₂ emissions with a total contribution of 9.3%. The largest contributing subcategory for point sources is the elemental phosphorous industry.
Nitrogen Oxides (NO\textsubscript{X})

The combination of all vehicles and motorized equipment is the greatest source of emissions of NO\textsubscript{X}, with 64.3% of the total source contribution. The largest subcategories for these contributors for NO\textsubscript{X} are off-road vehicles and equipment, and on-road gasoline vehicles. The small area source facilities (<100 tpy of any criteria pollutant) make up the bulk of the remaining contribution with 20%.
Volatile Organic Compounds

VOCs are included in the CAP emission inventory discussion as they are a primary precursor for ozone and a significant precursor of PM$_{2.5}$, criteria pollutants. The graph below shows that area source asphalt paving applications contribute the greatest portion of VOCs to the atmosphere with 39.9% of the total. All vehicles and equipment also contribute a significant amount, with 32.5% of the total. The largest contributing subcategories for area sources are asphalt applications, part degreasing, indoor wood burning, and surface coating (mainly furniture and vehicle).
Air Quality Standards

The federal Clean Air Act (CAA), last amended in 1990, requires EPA to set National Ambient Air Quality Standards (NAAQS) for pollutants considered harmful to public health and the environment. The standards are designed to primarily protect the general public, including sensitive populations such as asthmatics, children, and the elderly. They are also intended to safeguard public welfare by reducing effects such as decreased visibility and damage to animals, crops, vegetation, and buildings. EPA has established standards for six criteria pollutants. The list below contains seven pollutants, which include two size ranges of particulate matter.

The state of Idaho has adopted these standards. For more information, the US EPA air quality standards and supporting rationale are available at http://epa.gov/air/criteria.html.

EPA is also currently in the process of potentially revising the fine particulate NAAQS. Preliminary recommendations for a new standard propose daily and annual averages lower than current federal standards. (See http://www.epa.gov/ttn/naaqs/standards/pm/data/pmstaffpaper_20050630.pdf and http://www.epa.gov/sab/pdf/casac_pmp_mtg_april_6-7_2005_2nd_draft_pm_staff_paper-ra_draft_report_v2.pdf). Based on these recommendations, EPA has proposed a lower fine particulate NAAQS, a NAAQS for PMcoarse (PM$_{10-2.5}$), and is proposing to revoke PM$_{10}$.

### Air Quality Standards for Criteria Pollutants

<table>
<thead>
<tr>
<th>Pollutant</th>
<th>Standard</th>
<th>Level</th>
</tr>
</thead>
<tbody>
<tr>
<td>Ozone</td>
<td>The 3-year average of the 4th highest daily maximum 8-hour average concentration cannot exceed the level measured at each monitor within an area over each year.</td>
<td>0.084 ppm</td>
</tr>
<tr>
<td>Particulate Matter (10 micrometers)</td>
<td>The 3-year annual average of the weighted annual mean concentration at each monitor within an area must not exceed the level</td>
<td>54 µg/m$^3$</td>
</tr>
<tr>
<td></td>
<td>The 24-hour average cannot exceed the level more than once per year</td>
<td>154 µg/m$^3$</td>
</tr>
<tr>
<td>Particulate Matter (2.5 micrometers)</td>
<td>The 3-year annual average of the weighted annual mean concentrations cannot exceed the level</td>
<td>15.4 µg/m$^3$</td>
</tr>
<tr>
<td></td>
<td>The 3-year average of the 98th percentile (based on the number of samples taken) of the daily concentrations must not exceed the level</td>
<td>65 µg/m$^3$</td>
</tr>
<tr>
<td>Carbon Monoxide</td>
<td>The 1-hour average cannot exceed the level more than once per year</td>
<td>35 ppm</td>
</tr>
<tr>
<td></td>
<td>The 8-hour average cannot exceed the level more than once per year</td>
<td>9.4 ppm</td>
</tr>
<tr>
<td>Sulfur Dioxide</td>
<td>Annual arithmetic mean of 1-hour averages cannot exceed the level</td>
<td>0.03 ppm</td>
</tr>
<tr>
<td></td>
<td>24-hour average cannot exceed the level more than once per year</td>
<td>0.14 ppm</td>
</tr>
<tr>
<td></td>
<td>The 3-hour average cannot exceed the level more than once per year</td>
<td>0.50 ppm</td>
</tr>
<tr>
<td>Lead</td>
<td>The quarterly average (by calendar) cannot exceed the level</td>
<td>1.5 µg/m$^3$</td>
</tr>
<tr>
<td>Nitrogen Dioxide</td>
<td>The annual mean of 1-hour averages cannot exceed the level</td>
<td>0.053 ppm</td>
</tr>
</tbody>
</table>

Note: Daily concentration is the 24-hour average, measured from midnight to midnight.
In some instances, comparison of numbers in this table with sources listed above may appear to be slightly off (for example, ozone 8-hour standard 0.084 ppm listed in the table versus 0.08 ppm on the EPA Web site). These slight differences are due to a rounding convention adopted by EPA and the number of significant figures. The numbers shown on the table above are those used to determine if an area is in compliance, and are reflected in the graphs on the following pages.

The NAAQS for each pollutant may have different averaging times (for example, hourly and 8-hour averages). These different forms of the standard are created and enforced to address varied health impacts that happen as a result of a shorter, high-level exposure versus longer, low-level exposures. These differences are addressed pollutant-by-pollutant in the following sections, and additional information is on the EPA Web site listed above. A distinction exists between “exceeding” and “violating” a standard; the two are not equivalent. This distinction is due to the nature of the standards. In most instances it is allowable for an area to exceed the standard a few times, to allow for possible meteorological aberrances. For example, a carbon monoxide 8-hour average of 10 ppm clearly exceeds the standard; however it does not violate the standard if it is the only exceedance that year (the standard allows for one exceedance).

The EPA standards typically apply to an ‘area’, which may be defined in different ways. Data are often presented for individual monitoring stations in the following sections because this provides more insight into how pollutants are distributed in Idaho. The summaries that follow show how Idaho’s airsheds compared to the standards discussed above for the year 2004 and in many instances also incorporate the AQI and other measures of air quality where appropriate. The AQI color code shading is shown to aid interpretation of air quality, but does not imply whether or not standards were actually met for each pollutant; only meeting the conditions listed in the table above warrant compliance.
Ozone

Ozone is a summertime air pollution problem and is not directly emitted by pollutant sources. It forms when photochemical pollutants from cars and industrial sources (paints, solvents, gas vapors) react with sunlight. These pollutants are called ozone precursors and include VOC and NO\textsubscript{x}. Ozone levels are usually highest in the afternoon because of the intense sunlight and the time required for ozone to form. Ozone levels are highly affected by weather. DEQ monitored ozone from April through October in 2004, as this is the time period of concern for high ozone levels.

People frequently hear of ozone in the atmosphere. In this context, ozone is considered beneficial because it helps to protect the earth from the sun’s rays. In contrast, ozone formed at ground level is unhealthy. Elevated concentrations of ground-level ozone can cause reduced lung function and respiratory irritation, and can aggravate asthma. Ozone has also been linked to immune system effects (http://www.epa.gov/ttn/oarpg/naaqsfin/o3health.html). The damage ozone causes to the lungs heals within a few days, but repeated or prolonged exposure may cause permanent damage. People with respiratory conditions should limit outdoor exertion if ozone levels are high. Even healthy individuals may experience respiratory symptoms on a high-ozone day. Ground-level ozone can also damage agricultural crops and forests, interfering with their ability to produce food and grow.

The monitoring stations measuring ozone are located in both urban and rural areas, although the precursor chemicals that react with sunlight to produce ozone are generated primarily in large metropolitan areas. Because summers in southwest Idaho are normally hot and dry, the Treasure Valley tends to see daily ozone levels that begin to rise in the late morning and then peak in the late afternoon and early evening. This phenomenon follows very closely with the time of day that the temperatures are the hottest and the sun is the highest in the sky.

Graphs presented on the following pages show trends in ozone levels in the Treasure Valley airshed, reflecting both the AQI and the NAAQS. The graph on page 30 presents daily maximum 8-hour average data for the months of April through October, as these are the months where ozone levels are greatest. The shading on the graph corresponds to the AQI breakpoints for ozone, which is typically based on the 8-hour average.

The graph on page 31 shows monitoring data for each monitoring station against the federal standard, and shows that the region has remained below the standard since monitoring began. This means the 3-year average of the 4th-highest 8-hour concentration has not violated the NAAQS standard of 0.084 ppm. The ozone standard is defined such that the three highest concentrations can exceed the level of the standard while still maintaining attainment. Values presented on the graph are 3-year averages (of 4\textsuperscript{th}-highest concentrations); the year on the x-axis represents the last year averaged. For example, concentrations shown for 2004 are an average of 2002, 2003, and 2004 concentrations.

For additional information on ozone, visit www.epa.gov/air/urbanair/ozone/index.html. There is also additional information on ozone in question/answer format in the definitions section of this document.
Ozone (O₃) in the Treasure Valley
Daily Maximum 8-Hour Concentration
Treasure Valley 8-Hour Ozone
3-Year Average of 4th Highest Annual Concentration vs Standard

Year

8-Hour Average Concentration Parts per Million (ppm)

2003 2004

8-Hour Federal Standard
Whitney
Middleton
Tilli
Particulate Matter (10 micrometers)

Particulate matter (PM) includes both solid matter and liquid droplets suspended in the air. Particles smaller than 2.5 micrometers in diameter are called “fine” particles, or PM_{2.5}. Particles between 2.5 and 10 micrometers in diameter are called “coarse” particles. PM_{10} includes both fine and coarse particles. Coarse particles typically come from crushing or grinding operations and dust from roads. PM_{10} can aggravate respiratory conditions such as asthma. People with respiratory conditions should avoid outdoor exertion if PM_{10} levels are high.

The federal PM_{10} standard is currently being reviewed along with the fine particulate standard. EPA may choose to change PM_{10} in the future to include only the coarse particles, ranging from diameters 2.5-10 micrometers. In 2004 Idaho had three areas that had exceeded federal PM_{10} standards. These Nonattainment areas are in Sandpoint, Pinehurst and Pocatello. Northern Ada County was formerly a Nonattainment area but is now considered to be a Maintenance area for PM_{10}.

PM_{10} is monitored in Idaho using both reference and continuous methods. The PM_{10} TEOM is a federal equivalent method (FEM). TEOM data is also used to determine compliance to the PM_{10} NAAQS. Reference and FEM method results are shown in the following graphs. TEOM data is also used to inform the public of air quality values in near real time.

The graphs on the following pages demonstrate that Idaho’s airsheds were in compliance for both the annual and daily NAAQS standards for PM_{10}. Most concentrations were measured using the reference method while Sandpoint, Pinehurst, Coeur d’Alene, the Boise Fire Station, and Nampa were measured with TEOMs, the federal equivalent method. The graphs on pages 33 and 34 show PM_{10} at individual monitoring stations for each county. It is clear that all concentrations are below the NAAQS. Years shown on the x-axis are actually the last year that was averaged. For example, data points for 2004 are actually data points reflecting the 3-year average of 2002, 2003, and 2004.

The annual standard for 24-hour PM_{10} requires that the 3-year average not exceed the standard of 54 µg/m^3. Again, years shown on the x-axis are actually the last year that was averaged, as the standard requires a 3-year average. Maximum daily values (24-hour average) confirm that Idaho has shown a fairly slight decrease since 1995. Statistical summaries of reference and continuous method PM_{10} concentrations are provided in tables in the Appendix. The maximum PM_{10} measured in 2004 was 102 µg/m^3 at the Coeur d’Alene Lakes Middle School monitor, measured by a continuous equivalent method analyzer. This value is well below the 24 hour standard allowing each monitor one value above 154 µg/m^3 per year. (The 24-hour daily PM NAAQS is violated if the expected number of exceedances over three consecutive years is greater than one. So you can have 2/0/1 exceedances in three years and not violate the NAAQS.)

For additional information on PM_{10}, visit [www.epa.gov/air/urbanair/pm/index.html](http://www.epa.gov/air/urbanair/pm/index.html). More information on PM_{10} is also presented in question/answer format in the definitions section of this document.
Daily PM$_{10}$ 1995 - 2004
3-Year Average of Daily Maximum vs Standard Reference Method

[Graph showing 3-year average daily maximum concentration micrograms per cubic meter (μg/m$^3$) from 1995 to 2004.]
Annual PM$_{10}$ 1995 - 2004

Annual Mean vs Standard

3-Year Average Daily Concentration
Micrograms per Cubic Meter (ug/m$^3$)

Year


Federal Standard
SPT
CDA
PIN
LEW
BOI
Poc STP
IDF
Poc G&G
Particulate Matter (2.5 micrometers)

Particles smaller than 2.5 micrometers in diameter are called “fine” particles, or PM$_{2.5}$. DEQ considers PM$_{2.5}$ one of the major air pollution concerns affecting a number of airsheds in Idaho. PM$_{2.5}$ generally comes from wood burning, agricultural burning and other area sources, as well as vehicle exhaust including cars, diesel trucks, and buses. Fine particulate can also be formed secondarily in the atmosphere by chemical reactions of pollutant gases.

Exposure to PM$_{2.5}$ can have serious health effects. Fine particles are most closely associated with increased respiratory disease, decreased lung function, and even premature death. Children, older adults, and people with some illnesses are more sensitive and more likely to develop heart or lung problems associated with PM$_{2.5}$. People with respiratory or heart disease, older adults, and children should avoid outdoor exertion if PM$_{2.5}$ levels are high. PM$_{2.5}$ also significantly affects visibility.

PM$_{2.5}$ is primarily measured using two different methods in Idaho, the Federal Reference Method (FRM) and the Tapered Element Oscillating Method (TEOM). The federal reference method (FRM) is considered by EPA to be the most accurate way to determine PM$_{2.5}$ concentrations. This method involves pulling in air (at a given flow rate) and trapping particles of a certain size (in this case PM$_{2.5}$) on a filter. The filter is then weighed and divided by volume (determined from flow rate and amount of time) to provide concentration. Unfortunately, the FRM method does not provide continuous or timely information. Thus, Idaho uses another method to provide more time-relevant data: the Tapered Element Oscillating Microbalance (TEOM). The TEOM method uses measurement of mass to determine particulate matter present. A third method of PM$_{2.5}$ measurement is used during agricultural burning season; the nephelometer. These instruments help DEQ estimate PM$_{2.5}$ concentrations during these activities.

The graphs in this section use data primarily from the federal reference method (FRM). The continuous data are from the TEOMs. Every year all of the continuous methods are compared to the reference method values and calculations are made to determine the degree of difference from the reference method. The differences are then applied to the current continuous values in an attempt to make them “FRM -like.” Continuous methods are not designated as equivalent methods for PM$_{2.5}$ and data cannot be used for NAAQS determinations.

The graphs on pages 37 through 38 show that PM$_{2.5}$ meets both annual and daily NAAQS using the standard reference method. All of Idaho was designated attainment/unclassifiable for PM$_{2.5}$ in 2004. The graphs on pages 39 through 45 use data from the continuous samplers to display PM$_{2.5}$ with the AQI. Both types of graphs are discussed below.

The graph on page 37 shows the 2004 98$^{th}$ percentile 24-hour (daily) averages at each monitoring station against the federal standard. As shown in the standards table, the 98$^{th}$ percentile is actually for a 3-year average. For purposes of these graphs, however, the 3-year average is not taken because there are so few years to show (there would be only two or three points on each graph). It is easy to see, however, that data for all years fell well below the standard of 65 µg/m$^3$, and the 3-year average would also fall below.
The graph on page 38 shows annual averages at each monitoring station, against the federal standard. The standard calls for a 3-year average; however, annual averages are plotted on these graphs because 3-year averages would result in only a few data points. Nonetheless, it is easily seen that the annual standard of 15.4 $\mu g/m^3$ was not exceeded at any of the monitoring stations (nor would a 3-year average).

Graphs on pages 39 through 45 show daily PM$_{2.5}$ concentrations measured at Idaho sites during 2004 using the TEOM continuous analyzers against a backdrop of AQI breakpoints. The highest concentration of PM$_{2.5}$ measured in 2004 was 59 $\mu g/m^3$, measured at Idaho Falls using the FRM.

For additional information on particulate matter, visit [www.epa.gov/air/urbanair/pm/index.html](http://www.epa.gov/air/urbanair/pm/index.html). Information on PM$_{2.5}$ is also presented in a question/answer format in the definitions section of this document.
Daily PM$_{2.5}$ 2000 - 2004

98th Percentile vs Standard Reference Method

Year

2000 2001 2002 2003 2004

98th Percentile of Daily Concentrations Micrograms per Cubic Meter (ug/m$^3$)

0 10 20 30 40 50 60 70

Federal Standard
Boise
POC
SndPt
IDF
Nampa
CDA
Lew
PIN
TWF
Annual PM$_{2.5}$ 2000 - 2004
Annual Mean vs Standard Reference Method

Annual Average Daily Concentration
Micrograms per Cubic Meter (μg/m$^3$)

Year
2000 2001 2002 2003 2004

Federal Standard
Boise
POC
SndPt
IDF
Nampa
CDA
Lew
PIN
TWF
Treasure Valley
PM$_{2.5}$ Daily Averages from Continuous Analyzers

24-Hour Average Concentration
Micrograms per Cubic Meter

Jan Feb Mar Apr May Jun Jul Aug Sep Oct Nov Dec 2004

- >65.4 µg/m³ Unhealthy
- 40.5 - 65.4 µg/m³ Unhealthy for Sensitive Groups
- 15.5 - 40.4 µg/m³ Moderate
- 0 - 15.4 µg/m³ Good

Mass Transducer R&P TEOM 1400a - MtnView
Mass Transducer R&P TEOM 1400a - NNU
Coeur d'Alene
PM$_{2.5}$ Daily Averages from Continuous Analyzers

![Graph showing PM$_{2.5}$ daily averages from continuous analyzers in Coeur d'Alene, 2004.](chart)

- **24-Hour Average Concentration**: Micrograms per Cubic Meter
- **Legend**:
  - >65.4 µg/m³ Unhealthy
  - 40.5 - 65.4 µg/m³ Unhealthy for Sensitive Groups
  - 15.5 - 40.4 µg/m³ Moderate
  - 0 - 15.4 µg/m³ Good
  - Mass Transducer R&P TEOM 1400a - LMS
  - Mass Transducer R&P TEOM 1400a - Post Falls
Northern Idaho
PM$_{2.5}$ Daily Averages from Continuous Analyzers

24-Hour Average Concentration (Midnight-to-Midnight) Micrograms per Cubic Meter

- >65.4 µg/m³ Unhealthy
- 40.5 - 65.4 µg/m³ Unhealthy for Sensitive Groups
- 15.5 - 40.4 µg/m³ Moderate
- 0 - 15.4 µg/m³ Good

Mass Transducer R&P TEOM 1400a - Sandpoint
Mass Transducer R&P TEOM 1400a - Pinehurst
North Central Idaho
PM$_{2.5}$ Daily Averages from Continuous Analyzers

- >65.4 µg/m³ Unhealthy
- 40.5 - 65.4 µg/m³ Unhealthy for Sensitive Groups
- 15.5 - 40.4 µg/m³ Moderate
- 0 - 15.4 µg/m³ Good

- Mass Transducer R&P TEOM 1400a - Lewiston
- Mass Transducer R&P TEOM 1400a - Moscow
Twin Falls
PM$_{2.5}$ Daily Averages from Continuous Analyzers

2004 Air Quality Data Summary
Southeast Idaho
PM$_{2.5}$ Daily Averages from Continuous Analyzers

24-Hour Average Concentration
Micrograms per Cubic Meter

- >65.4 µg/m$^3$ Unhealthy
- 40.5 - 65.4 µg/m$^3$ Unhealthy for Sensitive Groups
- 15.5 - 40.4 µg/m$^3$ Moderate
- 0 - 15.4 µg/m$^3$ Good
- Mass Transducer R&P TEOM 1400a - Pocatello
- Mass Transducer R&P TEOM 1400a - S Springs
Eastern Idaho
PM$_{2.5}$ Daily Averages from Continuous Analyzers

![Graph showing PM$_{2.5}$ daily averages from continuous analyzers for Eastern Idaho in 2004.](image)

- >65.4 µg/m³ Unhealthy
- 40.5 - 65.4 µg/m³ Unhealthy for Sensitive Groups
- 15.5 - 40.4 µg/m³ Moderate
- 0 - 15.4 µg/m³ Good

Mass Transducer R&P TEOM 1400a - Idaho Falls
Mass Transducer R&P TEOM 1400a - Salmon
Carbon Monoxide

Carbon Monoxide (CO) is an odorless, colorless gas that can enter the bloodstream through the lungs and reduce the amount of oxygen that reaches organs and tissues. Carbon monoxide forms when the carbon in fuels doesn’t burn completely. The majority of CO comes from vehicle exhaust. In cities, 85-95% of all CO emissions may come from motor vehicle exhaust.

Elevated levels of CO in the ambient air can occur in areas with heavy traffic congestion. The highest levels of CO in the outside air typically occur during the colder months of the year when temperature inversions are more frequent. People with cardiovascular disease or respiratory problems might experience chest pain and increased cardiovascular symptoms, particularly while exercising, if CO levels are high. High levels of CO can affect alertness and vision even in healthy individuals.

CO monitoring stations are generally located in areas with heavy traffic congestion. These include central business areas, roadsides, and shopping malls.

The graph on page 47 shows the second highest 8-hour concentrations at Idaho’s monitoring sites versus the NAAQS standard. The second-highest concentration is displayed on these graphs because, under the federal rule, the 8-hour average can not be exceeded more than once per year (thus, choosing the second highest). These graphs confirm the general downward trend that CO is taking from the early 1990s to present. There were no 8-hour concentrations measured at any sites that exceeded the NAAQS standard of 9.4 ppm. The maximum 8-hour concentration for CO in 2004 was 3.4 ppm, well below the 8-hour standard. These data are provided in the Appendix.

The NAAQS also includes a 1-hour standard for CO of 35 ppm (can not be exceeded more than once a year). Measured 1-hour concentrations in Idaho are historically much lower than the 35 ppm standard, and therefore 1-hour CO trends were not graphed. The maximum and second-highest measured 1-hour CO in 2004 are 6.8 and 5.7 ppm. Additional 1-hour average CO data are provided in the Appendix.

For additional information on CO, visit www.epa.gov/air/urbanair/co/index.html. CO information is also provided in question/answer format in the definitions section of this document.
Carbon Monoxide (CO) for Idaho
2nd Highest 8-Hour Concentration vs Standard

<table>
<thead>
<tr>
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<tr>
<td>Boise</td>
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</tbody>
</table>

Lewiston sampling stopped in 2002
Sulfur Dioxide

Sulfur dioxide (SO$_2$) is a colorless, reactive gas produced by burning fuels containing sulfur, such as coal and oil, and by industrial processes. Historically, the greatest sources of SO$_2$ were industrial facilities that derived their products from raw materials like metallic ore, coal, and crude oil, or that burned coal or oil to produce process heat (petroleum refineries, cement manufacturing, and metal processing facilities). Currently, on-road vehicles, marine craft, and diesel construction equipment also release significant SO$_2$ emissions to the air.

People with asthma who are active outdoors may experience bronchoconstriction, where symptoms include wheezing, shortness of breath, and tightening of the chest. People should limit outdoor exertion if SO$_2$ levels are high.

The maximum measured SO$_2$ concentrations in 2004 were significantly below the federal standards. The graphs on pages 50 and 51 show the maximum 24-hour and 1-hour concentrations, respectively, at Idaho’s monitoring sites. The maximum 24-hour and 1-hour averages were 0.023 ppm and 0.081 ppm respectively.

Additional SO$_2$ data are located in the Appendix, and information on SO$_2$ is available at [www.epa.gov/air/urbanair/so2/index.html](http://www.epa.gov/air/urbanair/so2/index.html). SO$_2$ information is also provided in question/answer format in the definitions section of this document.
Sulfur Dioxide (SO₂)
Annual Average vs Standard

Maximum 1-Hour Average Concentration
Parts per Million (ppm)

- Annual Federal Standard
- Pocatello
- Soda Springs

Year
1999 2000 2001 2002 2003 2004
Sulfur Dioxide (SO₂)
Maximum 24-Hour Average vs Standard

Year
1999 2000 2001 2002 2003 2004

Maximum 24-Hour Average Concentration
Parts per Million (ppm)

- 24-Hour Federal Standard
- Pocatello
- Soda Springs
Sulfur Dioxide (SO₂)
Maximum 3-Hour Average vs Standard

<table>
<thead>
<tr>
<th>Year</th>
<th>Pocatello</th>
<th>Soda Springs</th>
</tr>
</thead>
<tbody>
<tr>
<td>1999</td>
<td>0.12</td>
<td>0.10</td>
</tr>
<tr>
<td>2000</td>
<td>0.11</td>
<td>0.09</td>
</tr>
<tr>
<td>2001</td>
<td>0.10</td>
<td>0.08</td>
</tr>
<tr>
<td>2002</td>
<td>0.10</td>
<td>0.08</td>
</tr>
<tr>
<td>2003</td>
<td>0.09</td>
<td>0.07</td>
</tr>
<tr>
<td>2004</td>
<td>0.08</td>
<td>0.06</td>
</tr>
</tbody>
</table>

3-Hour Federal Standard
Lead

Lead is a highly toxic metal that was used for many years in household products, automobile fuel, and industrial chemicals. Airborne lead was associated primarily with automobile exhaust and lead smelters. The large reductions in lead emissions from motor vehicles have changed the nature of the air quality lead problem in the United States. Industrial processes, particularly primary and secondary lead smelters and battery manufacturers, are now responsible for most of the lead emissions.

People, animals, and fish are mainly exposed to lead by breathing and ingesting it in food, water, soil, or dust. Lead accumulates in the blood, bones, muscles, and fat. Infants and young children are especially sensitive to even low levels of lead. Lead can have health effects ranging from behavioral problems and learning disabilities to seizures and death.

According to EPA, the primary sources of lead exposure are lead-based paint, lead-contaminated dust, and lead-contaminated residual soils. Refer to the EPA Web site www.epa.gov/ttnatw01/hlthef/lead.html for ways to limit your exposure to these lead sources.

Lead has not been monitored in Idaho since 2002. With the phase-out of lead in fuel and the closure of the Bunker Hill lead smelter in Kellogg, airborne lead is no longer a public health concern in Idaho. The graph on page 53 is included to show the historical monitoring of airborne lead in Kellogg.

For additional information on lead, visit www.epa.gov/air/urbanair/lead/index.html. Lead information is also available in a question/answer format in the definitions section of this document.
Lead (Pb)
Maximum Quarterly Average vs Standard

Maximum Quarterly Concentration
Micrograms per Cubic Meter (μg/m³)

Quarterly Federal Standard
Kellogg Medical Center

sampling stopped in 2002

Year

Nitrogen Dioxide

Nitrogen dioxide (NO₂) is a reddish brown, highly reactive gas that forms from the reaction of nitrogen oxide (NO) and oxygen in the atmosphere. The term “NOₓ”, which is frequently used, refers to both NO and NO₂. NO₂ will react with VOCs and can result in the formation of ozone. On-road vehicles like trucks and automobiles are the major sources of NOₓ. Industrial boilers and processes, home heaters, and gas stoves can also produce NOₓ. NO₂ pollution is greatest in cold weather.

NO₂ can cause respiratory symptoms such as coughing, wheezing, and shortness of breath in people with respiratory diseases such as asthma. Long-term exposure can lead to respiratory infections.

Motor vehicle manufacturers have been required to reduce NOₓ emissions from cars and trucks since the 1970s. NOₓ is not considered a significant pollution problem in Idaho. In 2004, DEQ only maintained two monitoring sites for nitrogen dioxide for the first part of the year at Mountain Home and established one at the end of the year near Coeur d’Alene. The monitoring objective is primarily to assess ambient NOₓ concentrations for evaluating ozone formation processes. The annual average for each year has consistently been less than half of the NAAQS standard, as shown in the graph on page 55 and in data in the Appendix. The maximum 1-hour average of NO₂ measured in 2004 was 0.048 ppm.

For additional information on NO₂, visit [www.epa.gov/air/urbanair/nox/index.html](http://www.epa.gov/air/urbanair/nox/index.html).
Idaho Nitrogen Dioxide (NO₂)
Annual 1-Hour Average vs Standard

Year

Annual 1-Hour Average Concentration
Parts per Million (ppm)

- Federal Standard
- Pocatello
- Boise
- Moyie Springs
- Mtn. Home

Nitrogen Dioxide
2004 Air Quality Data Summary
Visibility

There are no federal or state standards established for visibility. This parameter is presented (without comparison to a standard) as an easily-understood indicator of air quality. Visibility is often explained in terms of visual range and light extinction. **Visual range** is the maximum distance—usually miles or kilometers—that you can see a black object against the horizon. **Light extinction** is the sum of light scattering and light absorption by fine particles and gases in the atmosphere. The more light extinction you have, the shorter your visual range will be. Visual range as measured by nephelometer instruments utilizing light-scattering methodology provides an objective approach to measuring visibility at a specific location, but does not address individual perceptions regarding the “quality” of a view on a given day.

Reduced visibility (or visual range) is caused by weather (clouds, fog, and rain) and air pollution (fine particles and gases). The major pollution contributors are fine particulate matter (PM$_{2.5}$) emissions and photochemical smog, which are transported aloft and may remain suspended for a week or longer. Fine particles have a greater impact than coarse particles at locations far from the emitting source because they remain suspended in the atmosphere longer and travel farther. PM$_{2.5}$ also presents some of the most serious health hazards to the public, so you can roughly assume that the worse the visibility due to particulate matter, the unhealthier the air is to breathe.

For additional information on visibility, visit [http://www.epa.gov/air/visibility/index.html](http://www.epa.gov/air/visibility/index.html). Visibility information is also available in a question/answer format in the definitions section of this document.
## DEFINITIONS

### Calculation and Breakpoints for the Air Quality Index (AQI)

<table>
<thead>
<tr>
<th>Breakpoints for Criteria Pollutants</th>
<th>AQI Categories</th>
</tr>
</thead>
<tbody>
<tr>
<td>$O_3$ (ppm)</td>
<td></td>
</tr>
<tr>
<td>8-hour</td>
<td></td>
</tr>
<tr>
<td>0.000–0.064</td>
<td>Good (0–50)</td>
</tr>
<tr>
<td>0.065–0.084</td>
<td>Moderate (51–100)</td>
</tr>
<tr>
<td>0.085–0.104</td>
<td>Unhealthy for sensitive groups (101–150)</td>
</tr>
<tr>
<td>0.105–0.124</td>
<td>Unhealthy (151–200)</td>
</tr>
<tr>
<td>0.125–0.374</td>
<td>Very unhealthy (201–300)</td>
</tr>
<tr>
<td>(c) 0.405–0.504</td>
<td>Hazardous (401–500)</td>
</tr>
<tr>
<td>(c) 0.505–0.604</td>
<td></td>
</tr>
<tr>
<td>$O_3$ (ppm) 1-hour$^a$</td>
<td></td>
</tr>
<tr>
<td>0.000–0.064</td>
<td>Good (0–50)</td>
</tr>
<tr>
<td>0.065–0.084</td>
<td>Moderate (51–100)</td>
</tr>
<tr>
<td>0.085–0.104</td>
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<tr>
<td>(c) 0.405–0.504</td>
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</tr>
<tr>
<td>(c) 0.505–0.604</td>
<td></td>
</tr>
<tr>
<td>$PM_{2.5}$ (µg/m$^3$)</td>
<td></td>
</tr>
<tr>
<td>0.0–15.4</td>
<td>Good (0–50)</td>
</tr>
<tr>
<td>15.5–40.4</td>
<td>Moderate (51–100)</td>
</tr>
<tr>
<td>40.5–65.4</td>
<td>Unhealthy for sensitive groups (101–150)</td>
</tr>
<tr>
<td>65.5–100</td>
<td>Unhealthy (151–200)</td>
</tr>
<tr>
<td>100–150</td>
<td>Very unhealthy (201–300)</td>
</tr>
<tr>
<td>(c) 150–200</td>
<td>Hazardous (401–500)</td>
</tr>
<tr>
<td>$PM_{10}$ (µg/m$^3$)</td>
<td></td>
</tr>
<tr>
<td>0.0–4.4</td>
<td>Good (0–50)</td>
</tr>
<tr>
<td>4.5–9.4</td>
<td>Moderate (51–100)</td>
</tr>
<tr>
<td>9.5–12.4</td>
<td>Unhealthy for sensitive groups (101–150)</td>
</tr>
<tr>
<td>12.5–15.4</td>
<td>Unhealthy (151–200)</td>
</tr>
<tr>
<td>15.5–20.4</td>
<td>Very unhealthy (201–300)</td>
</tr>
<tr>
<td>(c) 20.5–30.4</td>
<td>Hazardous (401–500)</td>
</tr>
<tr>
<td>$CO$ (ppm)</td>
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<td>0.000–0.034</td>
<td>Good (0–50)</td>
</tr>
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<td>0.035–0.144</td>
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</tr>
<tr>
<td>0.145–0.224</td>
<td>Unhealthy for sensitive groups (101–150)</td>
</tr>
<tr>
<td>0.225–0.304</td>
<td>Unhealthy (151–200)</td>
</tr>
<tr>
<td>$SO_2$ (ppm)</td>
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<tr>
<td>0.000–4.4</td>
<td>Good (0–50)</td>
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<td>4.5–9.4</td>
<td>Moderate (51–100)</td>
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<td>9.5–12.4</td>
<td>Unhealthy for sensitive groups (101–150)</td>
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<tr>
<td>12.5–15.4</td>
<td>Unhealthy (151–200)</td>
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<tr>
<td>15.5–30.4</td>
<td>Very unhealthy (201–300)</td>
</tr>
<tr>
<td>$NO_2$ (ppm)</td>
<td></td>
</tr>
<tr>
<td>0.000–200</td>
<td>Good (0–50)</td>
</tr>
<tr>
<td>201–500</td>
<td>Moderate (51–100)</td>
</tr>
<tr>
<td>500–1000</td>
<td>Unhealthy for sensitive groups (101–150)</td>
</tr>
<tr>
<td>1000–2000</td>
<td>Unhealthy (151–200)</td>
</tr>
<tr>
<td>2000–5000</td>
<td>Very unhealthy (201–300)</td>
</tr>
<tr>
<td>(c) 5000–10000</td>
<td>Hazardous (401–500)</td>
</tr>
</tbody>
</table>

$^a$ Areas are generally required to report the AQI based on 8-hour ozone values. However, there are a small number of areas where an AQI based on 1-hour ozone values would be safer. In these cases, in addition to calculating the 8-hour ozone value, the 1-hour ozone value may be calculated, and the greater of the two values reported.

$^b$ $NO_2$ has no short-term National Ambient Air Quality Standard (NAAQS) and can generate an AQI only above a value of 200.

$^c$ 8-hour $O_3$ values do not define higher AQI values (above 300). AQI values above 300 are calculated with 1-hour $O_3$ concentrations.

For more detailed information about the AQI and the pollutants it measures, go to [www.epa.gov/airnow/aqibroch](http://www.epa.gov/airnow/aqibroch)
General Definitions

Air Toxics
Air toxics are broadly defined as almost 700 pollutants that DEQ considers potentially harmful to human health and the environment. These pollutants are listed in the Idaho’s air rules in IDAPA 58.01.01.585 and 586 (http://adm.idaho.gov/adminrules/rules/idapa58/0101.pdf). Hazardous air pollutants (see below) are included in this list to identify them as a subset of air toxics.

Criteria Air Pollutant (CAP)
The Clean Air Act of 1970 defined six criteria pollutants and established ambient concentrations to protect public health. EPA periodically has revised the original concentration limits and methods of measurement, most recently in 1997.

Hazardous Air Pollutant (HAP)
A hazardous air pollutant is an air contaminant identified as toxic in the Federal Clean Air Act, Section 112(b). 188 pollutants are currently listed as HAPs. They are listed by EPA at http://www.epa.gov/ttn/atw/188polls.html.

Temperature Inversions
The earth gains and loses most of its energy at its surface. It is warmed by solar heating during the day and cooled by radiation emissions at night. During the late morning and afternoon hours, the air near the surface is warmer than the air aloft and allows for good pollutant dispersion (vertical mixing may be 1,500 meters or more). At night with clear skies, the surface radiates heat into outer space, creating cooler air at the surface and warmer air aloft. Warmer air above cooler air (temperature inversion) is a stable condition and limits the upward movement of pollution because the warmer air acts as a barrier. With little or no wind, pollutants are trapped near the surface (vertical mixing may be 200 meters or less) and can reach high levels of concentration.

Volatile Organic Compound (VOC)
An organic compound that participates in atmospheric photochemical reactions. This excludes all compounds determined to have negligible photochemical reactivity by EPA and listed in 40 CFR 51.100(s) in effect July 1, 1998.

Visibility/Regional Haze
Visibility is often explained in terms of visual range and light extinction. Visual range is the maximum distance—usually miles or kilometers—that you can see a black object against the horizon. Light extinction is the sum of light scattering and light absorption by fine particles and gases in the atmosphere. The more light extinction you have, the shorter your visual range will be. Reduced visibility (or visual range) is caused by weather (clouds, fog, and rain) and air pollution (fine particles and gases). The major pollution contributor is fine particulate matter (PM$_{2.5}$) emissions, which are transported aloft and may remain suspended for a week or longer. Fine particles have a greater impact than coarse particles at locations far from the emitting source because they remain suspended in the atmosphere.
longer and travel farther. PM$_{2.5}$ also presents some of the most serious health hazards to the public, so you can roughly assume that the worse the visibility, the unhealthier the air is to breathe.

**Criteria Air Pollutants**

**Ozone (O$_3$)**

- **What is it?**
  Ozone, a bluish-colored gas molecule with a strong odor, is composed of three atoms of oxygen. In the upper atmosphere ozone occurs naturally and partially absorbs the sun’s harmful ultraviolet rays. Ozone at ground level is a summertime air pollution problem.

- **How is it caused?**
  Ozone forms when photochemical pollutants from cars, trucks, and industrial sources react with sunlight. Ozone-forming pollutants include NO$_X$ and VOCs. Even gasoline-powered yard equipment, paints, solvents, and off-road vehicle motors contribute.

- **When does it happen?**
  Ozone pollution is most common in the summer months, when sunlight and stable atmospheric conditions occur. Ozone levels are usually highest in the afternoon, as sunlight photochemically transforms NO$_X$ and VOCs into ozone.

- **Who is affected?**
  Adults and children who are active outdoors, people with respiratory disease such as asthma, and people with unusual sensitivity to ozone. During physical activity, ozone penetrates deeper into the lungs and can do more damage.

Ozone is a very reactive gas. For this reason, high concentrations of ozone can cause respiratory distress and disease in humans, decreased yields of agricultural crops and forests, and damage to some rubber products, plastics, and paints used outdoors. National crop losses from ozone exposure are estimated at $3 billion to $5 billion annually. Forest losses are harder to estimate.

- **What are the health effects?**
  Ozone can cause coughing and throat irritation, make deep vigorous breathing more difficult, and increase the chance of respiratory infections. It increases sensitivity to allergens and can trigger asthma attacks. The damage it causes to the lungs heals within a few days, but repeated or prolonged exposure may cause permanent damage.

- **What can I do about it?**
  If ozone levels are high and you have a respiratory condition or are normally active outdoors, try to limit your outdoor exertion.

In the United States, management of ozone and other photochemical oxidants has been a major goal of federal and state clean air legislation (Clean Air Act). Although many of the pollution control efforts required by the CAA have been implemented, efforts to decrease ozone pollution have been only partially successful.
In the Treasure Valley airshed the ozone trend is flat and is marginally within the Federal standards.

- **Where is it measured?**
  Unlike other pollutants monitored here in Idaho, ozone is formed by precursors that react in the atmosphere. Winds transport ozone and chemical emissions from one area to another. For the Treasure Valley, ozone precursors are emitted into the air in industrial areas of the airshed and subsequently travel southeasterly to more rural areas as they react to form ozone. As a result, for the Treasure Valley airshed, DEQ has monitors at both ends of the airshed and one in Boise. Another ozone monitor will run in the Coeur d’Alene area beginning in 2005.

**Particulate Matter (PM2.5 and PM10)**

- **What is it?**
  Particulate matter (PM) includes both solid matter and liquid droplets suspended in the air. Particles smaller than 2.5 micrometers in diameter are called “fine” particles, or $PM_{2.5}$. Particles between 2.5 and 10 micrometers in diameter are called “coarse” particles. $PM_{10}$ includes both fine and coarse particles. DEQ considers $PM_{2.5}$ one of the major air pollution concerns affecting our state.

- **How is it caused?**
  - $PM_{2.5}$ comes from all types of combustion, including cars, diesel trucks, power plants, and wood burning, and from some industrial processes. It can also be formed in the atmosphere by chemical reactions of pollutant gases.
  - The “coarse” particles in $PM_{10}$ typically come from crushing or grinding operations and dust from roads.

- **When does it happen?**
  Any time.

- **Who is affected?**
  People with asthma and heart or lung diseases, the elderly, and children. $PM_{2.5}$ also significantly affects visibility.

- **What are the health effects?**
  Fine particulates ($PM_{2.5}$) pose a greater risk to human health than coarse particulates, because they penetrate deeper into the respiratory system.
  - $PM_{2.5}$ exposure can have serious health effects. People with heart or lung diseases are at increased risk of attacks or premature death. Children and the elderly are more likely to develop heart or lung problems.
  - $PM_{10}$ can aggravate respiratory conditions such as asthma.

- **What can I do about it?**
  - If $PM_{2.5}$ levels are high, people with respiratory or heart disease, the elderly, and children should avoid outdoor exertion.
  - If $PM_{10}$ levels are high, people with respiratory conditions should avoid outdoor exertion.
Where is it measured?
Due to the health risks associated with PM, both PM$_{2.5}$ and PM$_{10}$ are monitored in various locations throughout Idaho.

Carbon Monoxide (CO)
- What is it?
  CO is an odorless, colorless gas that can enter the bloodstream through the lungs and reduce the amount of oxygen that reaches organs and tissues.
- How is it caused?
  Carbon monoxide forms when the carbon in fuels doesn’t burn completely. 60% of all CO comes from vehicle exhaust; and up to 95% in cities.
- When does it happen?
  CO pollution is at its worst in cold weather because fuels burn less efficiently in low temperatures. It usually peaks during morning and evening rush hours.
- Who is affected?
  People with cardiovascular disease, such as angina, or cardiovascular or respiratory problems; also possibly fetuses and young infants.
- What are the health effects?
  Chest pain and increased cardiovascular symptoms, particularly while exercising. High levels of CO can affect alertness and vision even in healthy individuals.
- What can I do about it?
  If CO levels are high, limit exertion and avoid sources of CO such as heavy traffic.
- Where is it measured?
  CO monitoring stations are located in areas with heavy traffic congestion. These include central business areas, roadsides, and shopping malls.

Sulfur Dioxide (SO$_2$)
- What is it?
  Sulfur dioxide is a colorless, reactive gas.
- How is it caused?
  SO$_2$ is produced by burning sulfur-containing fuels such as coal and oil, and by industrial processes.
- Where does it happen?
  The highest concentrations of SO$_2$ are usually near large industrial sources.
- Who is affected?
  People with asthma who are active outdoors.
- What are the health effects?
  Bronchoconstriction, which can cause wheezing, shortness of breath, and tightening of the chest. When exposure to SO$_2$ ends, the symptoms should clear up within an hour.
• **What can I do about it?**
  If SO₂ levels are high, limit your outdoor exertion.

• **Where is it measured?**
  Because the large primary sources of SO₂ in Idaho are industrial, DEQ monitors for SO₂ near large facilities with high SO₂ emissions. The only monitor running in 2004 was in Pocatello.

**Lead (Pb)**

• **What is it?**
  Lead is a highly toxic metal that was used for many years in household products, automobile fuel, and industrial chemicals.

• **How is it caused?**
  Locally, airborne lead is associated primarily with automobile exhaust and lead smelters. Since the phase-out of lead in fuels, however, cars and trucks are no longer a significant source of lead. Also, the Kellogg Bunker Hill Mine ceased operations in 1981.

• **When does it happen?**
  Lead concentrations are likely to be highest near sources where current or former lead smelting/processing operations caused particle fallout, especially in nearby soils such as unpaved parking lots.

• **Who is affected?**
  Everyone. Children 6 years and younger are most at risk.

• **What are the health effects?**
  Lead can have health effects ranging from behavioral problems and learning disabilities to seizures and death.

• **What can I do about it?**
  According to EPA, the primary sources of lead exposure are lead-based paint, lead-contaminated dust, and lead-contaminated residual soils. Refer to EPA’s Web site at [http://www.epa.gov/ttn/atw/hlthef/lead.html](http://www.epa.gov/ttn/atw/hlthef/lead.html) for ways to limit your exposure to these lead sources.

• **Where is it measured?**
  Due to the phase-out of leaded fuels and the closure of Idaho’s only lead smelter in 1981, DEQ no longer monitors for airborne lead. Historical monitoring was continued until 2002 but was discontinued due to the low levels being measured.

**Nitrogen Dioxide (NO₂)**

• **What is it?**
  Nitrogen dioxide (NO₂) is a reddish brown, highly reactive gas that forms from the reaction of nitrogen oxide (NO) and oxygen in the atmosphere. NO₂ will react with VOCs and can result in the formation of ozone.

• **How is it caused?**
  High temperature combustion sources such as power plants and automobiles are major producers of NO. Home heaters and gas stoves can also produce NO.
• **When does it happen?**
  NO\(_2\) pollution is greatest in cold weather. It follows a similar trend to CO.

• **Who is affected?**
  People with respiratory diseases such as asthma; also children.

• **What are the health effects?**
  NO\(_2\) can cause respiratory symptoms such as coughing, wheezing, and shortness of breath. Long-term exposure can lead to respiratory infections.

• **What can I do about it?**
  Since the 1970s, motor vehicle manufacturers have been required to reduce NO emissions from cars and trucks. It is not a significant pollution problem in Idaho.

• **Where is it measured?**
  Because NO\(_2\) is not a major concern in Idaho, it was measured during 2004 at only one location, Mountain Home. A second site will be collocated with an ozone monitor near Coeur d’Alene in 2005.

**Pollution Sources**

**Area Sources**
Countywide categories of pollution sources, in which each individual source emits pollutants below the thresholds for a point source facility.

**Biogenics**
Natural sources such as trees, plants, grass, crops, and soils. The worldwide emission rate of these natural hydrocarbons has been estimated to exceed that of non-methane hydrocarbons originating from human sources. Isoprene, one of the major constituents of biogenic emissions, is very photoreactive, and would seem to make biogenic VOC a contributor in the formation of ozone. The study of hydrocarbon emissions from plants is therefore of key importance to our understanding of the global effects of naturally produced hydrocarbons.

**Emission Factor**
A value derived from source tests, material balance calculations, or engineering comparisons with similar processes. Used to estimate emissions from process quantities.

**Non-road Mobile Sources**
Farm vehicles, on-site construction/industrial vehicles, logging equipment, small marine craft, aircraft, trains, lawn and garden equipment.

**On-road Mobile Sources**
Cars, trucks, sport utility vehicles, and buses.
**Point Sources**
For the every third year statewide emission inventory, point sources are defined as facilities that have annual air contaminant emissions equal to or exceeding 1000 tons per year of CO; 100 tons per year of nitrogen oxides (NO\textsubscript{X}), PM\textsubscript{10}, PM\textsubscript{2.5}, sulfur oxides (SO\textsubscript{x}) such as SO\textsubscript{2} and sulfur trioxide (SO\textsubscript{3}), or volatile organic compounds (VOC).

**Registered Facility**
The total of all pollutant-emitting activities located on adjacent or contiguous properties owned or operated by one person or corporate entity. It includes all of the pollutant-emitting buildings, processes, structures, equipment, control apparatuses, and storage areas at a facility.

**Stationary Area Sources**
Also called area sources. Pollution sources where each individual source emits pollutants below the thresholds for a point source facility. Sources include wood stoves/fireplaces, outdoor burning, architectural surface coating, automobile painting, commercial/consumer solvents, dry cleaning, printing, stationary diesel engines, and small utility engines, and construction activities.
2004 Air Quality Data Summary Appendix

December 2005
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### 2004 Air Quality Data Summary Appendix

**US EPA - AirData Monitor Values Report - Criteria Air Pollutants**
Friday, 21-Oct-2005 at 11:11:30 AM (USA Eastern time zone)

Geographic Area: Idaho
Pollutant: Particulate (2.5 micrometers)
Year: 2004

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US EPA - AirData Monitor Values Report - Criteria Air Pollutants
Friday, 21-Oct-2005 at 11:5:45 AM (USA Eastern time zone)
Geographic Area: Idaho
Pollutant: Ozone
Year: 2004

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US EPA - AirData Monitor Values Report - Criteria Air Pollutants  
Friday, 21-Oct-2005 at 10:20:38 AM (USA Eastern time zone)  
Geographic Area: Idaho  
Pollutant: Carbon Monoxide  
Year: 2004

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US EPA - AirData Monitor Values Report - Criteria Air Pollutants
Friday, 21-Oct-2005 at 11:3:56 AM (USA Eastern time zone)
Geographic Area: Idaho
Pollutant: Nitrogen Dioxide
Year: 2004

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### US EPA - AirData Monitor Values Report - Criteria Air Pollutants

#### Friday, 21-Oct-2005 at 11:7:46 AM (USA Eastern time zone)

**Geographic Area:** Idaho  
**Pollutant:** Sulfur Dioxide  
**Year:** 2004

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### US EPA - AirData Monitor Values Report - Criteria Air Pollutants

**Friday, 21-Oct-2005 at 11:25:31 AM (USA Eastern time zone)**

**Geographic Area:** Idaho  
**Pollutant:** Lead  
**Year:** 1999 - 2002

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